Structures of New Antibiotic Substances, Sakyomicin A, B, C, and D; X-Ray Crystal and Molecular Structure of Sakyomicin A

Hiroshi Irie,*a Yukio Mizuno,a Isao Kouno,a Toru Nagasawa,b Yoshiki Tani,b Hideaki Yamada,b Tooru Taga, and Kenji Osaki

^a Faculty of Pharmaceutical Sciences, Nagasaki University, Nagasaki, Japan

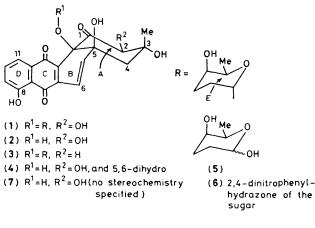
^b Department of Agricultural Chemistry, Kyoto University, Kyoto, Japan

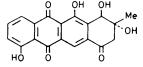
^c Faculty of Pharmaceutical Sciences, Kyoto University, Kyoto, Japan

The structure of a new antibiotic substance, sakyomicin A has been elucidated by X-ray crystallographic analysis and structures for its congeners, sakyomicin B, C, and D have been proposed from their spectroscopic properties.

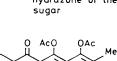
Investigation of cultures of Nocardia sp. No. 53 resulted in the isolation of four new antibiotic substances, sakyomicin A (1), B (2), C (3), and D (4) effective to Gram-positive bacteria. We report here the structure elucidation of these metabolites.

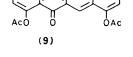
Compound (1), $C_{25}H_{26}O_{10}$, m.p. 205–207 °C (from n-hexane-acetone), $[\alpha]_{D}^{20}$ –99.4° (c 0.8, EtOH), λ_{max} (EtOH) (e) 216, 238, 310, and 415 nm (28,500, 15,900, 5,300, and 4,500), v_{max} (KBr) 3490, 3200, 1720, and 1638 cm⁻¹, and $\delta_{\rm H}$ (CD₃OD, 360 MHz) 0.56 (3H, d, J 6.6 Hz, 6'-Me) (see Figure 1 for numbering scheme used), 1.25 (3H, s, 13-Me), 1.63 and 1.93 [1H each, m, C(2')-H₂], 1.84 and 2.10 [1H each, broad d and m, C(3')-H₂], 2.00 and 2.12 [1H each, d, J 14.9 Hz, C(4)-H₂], 3.42 [1H, br. s, C(4')-H], 3.70 [1H, q, J 6.6 Hz, C(5')-H], 4.23 [1H, s, C(2)-H], 5.37 [1H, diffused d, J 2.8 Hz, C(1')-H], 6.46 [1H, d, J 10.0 Hz, C(5)-H], 6.93 [1H, d, J 10.0 Hz, C(6)-H], 7.37 [1H, d, J 7.9 Hz, C(9)-H], 7.65 [1H, d, J 7.9 Hz, C(11)-H], and 7.77 [1H, t, J 7.9 Hz, C(10)-H] gave (2) and a hexose (5) by hydrolysis with 2 M HCl at room temp. for 1 h. The latter was characterised as its 2,4-dinitrophenylhydrazone (6) ($M^+ = 312.1097$: $C_{12}H_{16}N_4O_6$ requires 312.1070), m.p. 117–118 °C, $[\alpha]_D^{26}$ $+14.8^{\circ}$ (c 0.5, pyridine), $\delta_{\rm H}$ (CD₃OD, 200 MHz) 1.18 (3H, d, J 6.2 Hz), 1.78 (2H, m), 2.56 (2H, m), 3.43 (1H, d-t, J 4.6 and 9.2 Hz), 3.64 (1H, d-q, J 4.6 and 6.2 Hz), 7.32 (1H, s), 7.79 (1H, t, J 5.2 Hz), 7.96 (1H, d, J 9.6 Hz), 8.31 (1H, d-d, J 9.6 and 2.8 Hz), and 9.01 (1H, d, J 2.8 Hz), and was proposed to be the enantiomer of (-)-rhodinose (2,3,6-trideoxy-











L-threo-hexose).¹ A single-crystal X-ray analysis of (1) was carried out.

Crystal data: orthorhombic, $P2_12_12_1$, a = 15.901(5), b =12.771(9), c = 11.232(3) Å, U = 2280.9 Å³, Z = 4, $D_x =$ 1.416 g/cm³. The structure was determined on a FACOM M200 computer, using 1845 reflection data collected on a Rigaku AFC-5 diffractometer with graphite-monochromated Mo- K_{α} radiation; current R = 0.054.[†] Figure 1 shows the structure of (1). The sugar group is located above the B- and c-rings, and the methyl group in the sugar moiety has a short contact with one of the keto-functions on the c-ring.

Based on the absolute configuration of the sugar (5), the absolute configuration of (1) was elucidated. Recently Ohta, Okazaki, and Kishi reported the structure of P-1894B (rineomycin A_1) based on the configuration of the sugar obtained from its hydrolysate.¹ The aglycone part of (1) is enantiomeric to that of P-1894B and, to our knowledge, (1) is the first naturally occurring compound containing (+)rhodinose.

The structure of (2), aglycone of (1), is similar to that of of yoronomycin (7),² but not identical since they differ in their $[\alpha]_{D}$ values [(2), 31.6°; yoronomycin, 73.5° in dioxan].

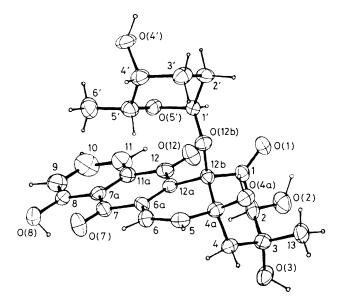


Figure 1. Structure of sakyomicin A (1) showing the crystallographic numbering system used.

† The atomic co-ordinates for this work are available on request from the Director of the Cambridge Crystallographic Data Centre, University Chemical Laboratory, Lensfield Road, Cam-bridge CB2 1EW. Any request should be accompanied by the full literature citation for this communication.

In addition, the ¹H n.m.r. spectrum of yoronomycin in $(CD_3)_2SO$ was measurable but no spectrum of (2) could be obtained in $(CD_3)_2SO$ owing to line broadening. Irradiation of (2) with sunlight in methanol resulted in formation of the tetracyclinone (8) in 67% yield by electrocyclic ring opening of the B-ring followed by tautomerisation and recyclisation. In 1977, the transformation of yoronomycin into the tetracyclinone (8') (no stereochemistry indicated) was reported³ but its ¹H n.m.r. spectrum was not identical with that of (8) by direct comparison.[‡] Acetylation of (8) (acetic anhydride-pyridine) gave the acetate (9), the i.r. spectrum of which was identical with that of the acetate obtained from (8');³ this suggests that yoronomycin is in fact a diastereoisomer of (2) at C(2) and/or C(3).

Sakyomicin C, $C_{25}H_{26}O_9$, m.p. 143—145 °C, $[\alpha]_D^{21} - 82.7^{\circ}$ (c 0.8, MeOH), λ_{max} (EtOH) (ϵ), 216, 240, 310, and 415 nm (20,000, 11,500, 3,500, and 4,400), ν_{max} (KBr) 3400, 1722, and 1635 cm⁻¹, showed signals at δ 2.46 [1H, d, J 11.8 Hz, C(2)-H] and 2.75 [1H, d-d, J 11.8 and 2.3 Hz, C(2)-H] in its ¹H n.m.r. spectrum and its ¹³C n.m.r. (CDCl₃) spectrum exhibited a signal at 55.78 p.p.m. (t). Based on its molecular formula and spectroscopic properties, the structure was proposed as (3).

[‡] We have, to date, been unable to obtain an authentic specimen for direct comparison.

The structure of sakyomicin D, $C_{19}H_{18}O_8$, m.p. 161–-163 °C (benzene solvate) $[\alpha]_{2^0}^{p_0}$ -140° (c 0.8, MeOH), λ_{max} (EtOH), (ϵ) 213, 249, 273, and 425 nm (37,000, 8,400, 8,800, and 3,600), v_{max} (KBr) 3400, 1728, and 1640 cm⁻¹ was proposed to be (4), since hydrogenation of (2) (5% Pd-carbon) gave (4) in good yield.

We thank Prof. N. Morimoto and Dr. K. Tomita, Institute of Geology and Mineralogy, Faculty of Sciences, Kyoto University for use of the AFC-5 X-ray diffractometer and Drs. H. Naoki and T. Iwashita, Suntory Institute for Bioorganic Research for measurement of the 360 MHz ¹H and ¹³C n.m.r. spectra.

Received, 20th September 1982; Com. 1111

References

- K. Ohta, H. Okazaki, and T. Kishi, *Chem. Pharm. Bull.*, 1982, 30, 43, and references therein; A. H. Haines, *Carbohydr. Res.*, 1972, 21, 99; C. L. Stevens, P. Blumbergs, and D. L. Wood, *J. Am. Chem. Soc.*, 1964, 86, 3592.
- 2 S. Matsumura, Y. Ezure, M. Ozaki, K. Kumagai, and H. Matsunaga, *Chem. Abs.*, 1977, 86, 104453s: Japan Kokai, 76-121,600, Oct., 1976.
- 3 S. Matsumura, M. Ozaki, K. Kumatani, and T. Matsunaga, *Chem. Abs.*, 1978, 88, 37507p; Japan Kokai, 77,111,554, Sept., 1977.